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TESTING THE RECYCLABILITY OF POLYMERS IN LABORATORY CONDITIONS

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Year 2022: Facts on plastics



- Plastics directly determine our reality both technically and ethically.
- Our standard of living is inconceivable without plastics.
- Plastics allowed the development of all technologies and activities of human life.
- Big effort is still made to the property development and production of plastics.
- Plastic waste is still not resolved so far (!).
- Consequences plastic waste contaminates nearly the whole planet.
- Only in the last decade has the problem of plastic disposal finally been media solved.

How does plastic differ from classic materials?



- Flexibility
- Easy processability
- Price
- Mass
- Resistance to corrosion
- Mechanical properties such as strength, ductility, toughness
- Barrier properties
- Dyeability
- Organoleptic properties

Plastics versus:

STONE WOOD METAL GLASS CERAMICS BIOPOLYMERS



Potential fate of plastics waste

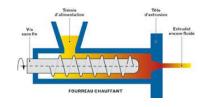






Outdoor deposits

Out of control





Landfill

a) controlledb) uncontrolled



Recycling

a) directb) chemical



Waste-to-Energy

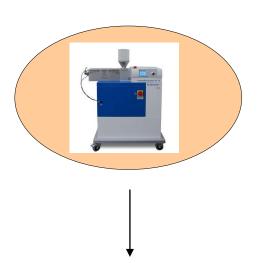
Professional

Laboratory approach X industrial approach



LABORATORY

Research facility



is needed

Co-operation

PRODUCTION

Plant production facility



Standard plastics converter

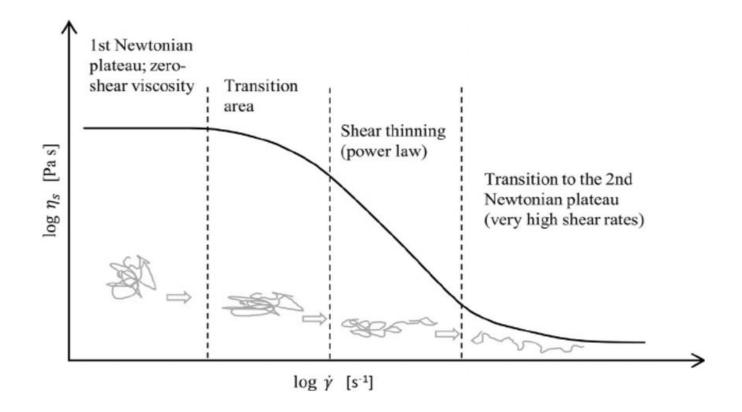
Company <u>specialized</u> to plastics recyclate processing

- a) Commercial research facility
- b) Academic research facility

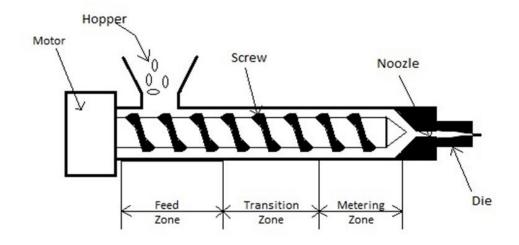
Rheology of polymer melt



- Viscoelastic properties determine behaviour during processing.
- Energy needed for polymer melting, its homogenization and shaping into the final geometry determines the extent of mechanochemical deterioration of polymer.

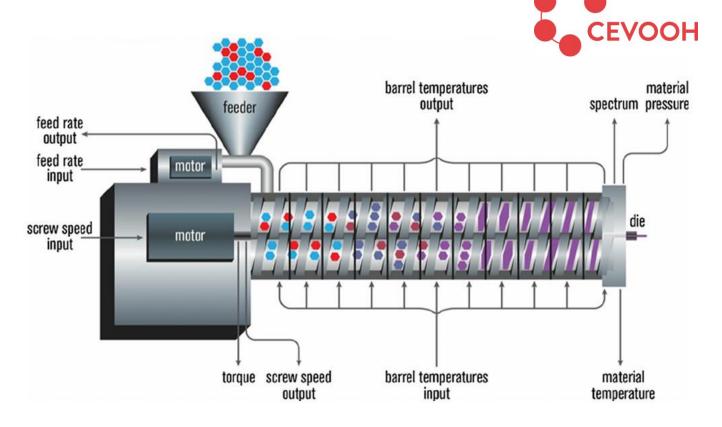


Types of extruders



Extruder (single-screw)

Basic part of most of polymer-processing technologies



Extruder (twin-screw)

Polymer compounding with additives of substantially different properties

Area of the highest mechanical stress

Specific features of plastics separated from municipal waste



- Variety of classes.
- Variety of types of the same polymer.
- Contamination.
- Degradative deterioration **processing**, environmental.

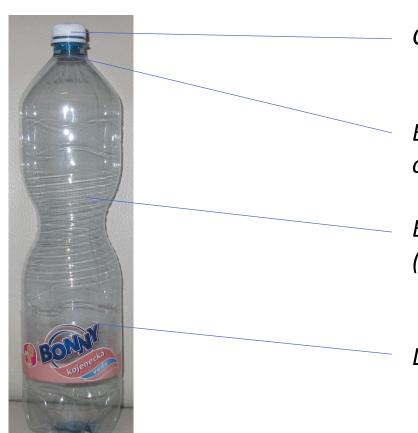






A case study – materials of drink bottle





Cap (HDPE)

Bottle neck and thread (**PET** crystalline)

Bottle body (**PET** partly amorphous)

Label (**PP**-biaxially oriented)

HDPE (high-density polyethylene)

PET (polyethylene glycol terephthalate)

PP (polypropylene)

to be tested in terms of recycling

Materials investigated

CEVOOH

- 1) PET-blue
- 2) PET-green
- 3) PET-transparent
- 4) HDPE-caps
- 5) HDPE-mix (ref.)
- 6) PP-labels

Processing stability by multiple extrusion.

Melt-index as a criterion of polymer deterioration.

Single-screw extruder HAAKE
Screw diameter 19 mm

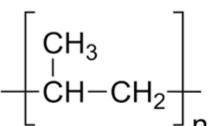
Length/diameter ratio = 25

Origin: Czech municipal waste.

Polymers separated within the terrain research carried out as a part of project TIRSMZP719 complying with the newly certified methodology.

PP

Chemický název: Polypropylen; Poly(1-methylethan-1,2-diyl)



Typ plastu: semikrystalický polymer

Skelný přechod: -10 °C Bod tání: 160-165 °C

Hustota: 0.905 g/cm³ (homopolymer)

Teplota rozkladu: 447 °C

PET/PETE

Chemický <u>název:</u> Polyethylen tereftalát; polyethylenglykol tereftalát

$$\begin{bmatrix} 0 & 0 \\ \parallel & \parallel \\ C - C - C - CH_2 - CH_2 - CH_2 - CH_2 \end{bmatrix}$$

Typ plastu: semikrystalický polymer

 Skelný přechod:
 70-80 °C

 Bod tání:
 250-260 °C

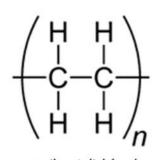
 Hustota:
 1.38 – 1.40 g/cm³

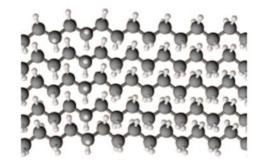
 Teplota rozkladu:
 425-440 °C



HDPE

Chemický <u>název:</u> Vysokohustotní polyethylen; lineární polyethylen





Typ plastu: semikrystalický polymer

Skelný přechod: -100 °C Bod tání: 125-138 °C

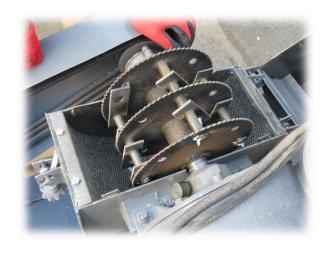
Hustota: HDPE 0.940 - 0.960 g/cm³ (high-density polyethylene)

UHMWPE...... 0.930 - 0.940 g/cm³ (ultra high-density polyethylene)

Teplota rozkladu: 485-495 °C

Processing









Grinder

Material after grinding

Granulate – the end product of polymer waste processing

The way from the flaky waste to granulate is not so easy as it looks!

Specific features of polymer waste laboratory processing



Playing a key role in success of laboratory testing

1. Physical form:

Has to comply with the cross section of hopper feeding area. Ground (or cut) polymer has to fall freely from the hopper down to the extruder screw.

In our case the hopper feeding area was the rounded rectanle 22 x 35 mm

Milling and grinding the polymer prior to extrusion is necessary!

Effect of heat exposure on PET bottle-flakes size



Flakes-cut polymer

200°C / 25 min.

250°C / 10 min.

Physical form unprocessable on laboratory equipment





Physical form – the final solution





Polymer ground and separated using 4 mm sieve **Laboratory processable**

Particles above 4 mm sieve **Laboratory non-acceptable**

Specific features of polymer waste laboratory processing



Playing a key role in success of laboratory testing

2. Contaminants:

- Other polymers
- Dust
- Mechanical contaminants
- Volatiles
- Foodstuffs
- Chemicals
- Biological pollutions

Manual separation and washing prior to extrusion is inevitable!

Paper bits (labels?)



Volatiles make the pellets spongy



Waste polymers for testing





HDPE-mix

HDPE-caps

PET-blue

PET-green

PET-transparent

Density HDPE (23 °C)

Density PET (23 °C)

 ρ = 0.94-0.96 g/cm³

 $\rho = 1.38-1.40 \text{ g/cm}^3$

PET-blue

CEVOOH

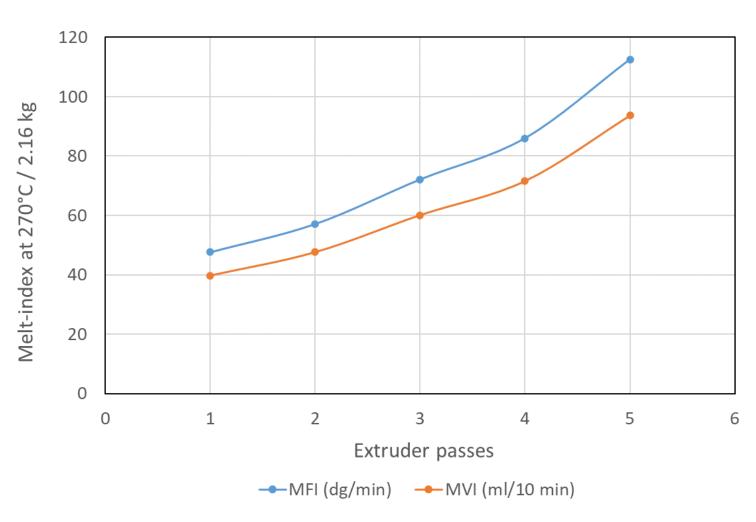
Processing stability determination

HAAKE single-screw extruder 19 mm, L/D=25

5-fold extrusion at 275°C/80 rpm

Before and between extrusions dried 3 hrs at 150 °C

Melt density 1.2 g/cm³ used for MFI calculation





Processing stability determination

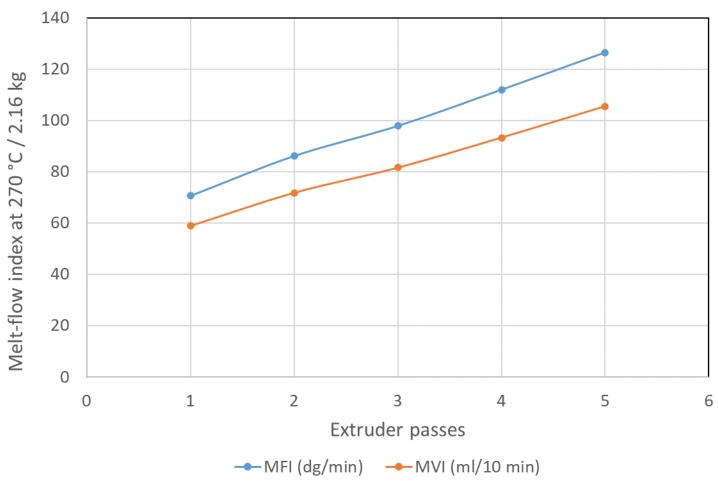
HAAKE single-screw extruder 19 mm, L/D=25

5-fold extrusion at 275°C/80 rpm

Before and between extrusions dried 3 hrs at 150 °C

Melt density 1.2 g/cm³ used for MFI calculation





PET-transparent

Processing stability determination

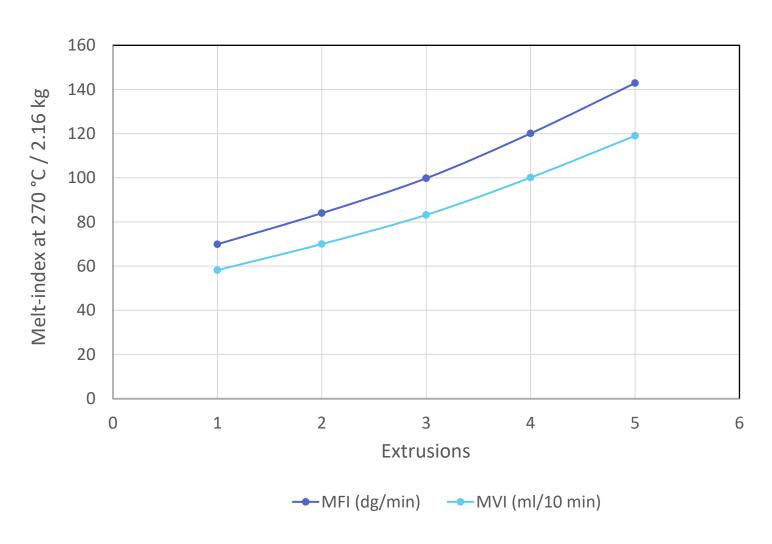
HAAKE single-screw extruder 19 mm, L/D=25

5-fold extrusion at 275°C/80 rpm

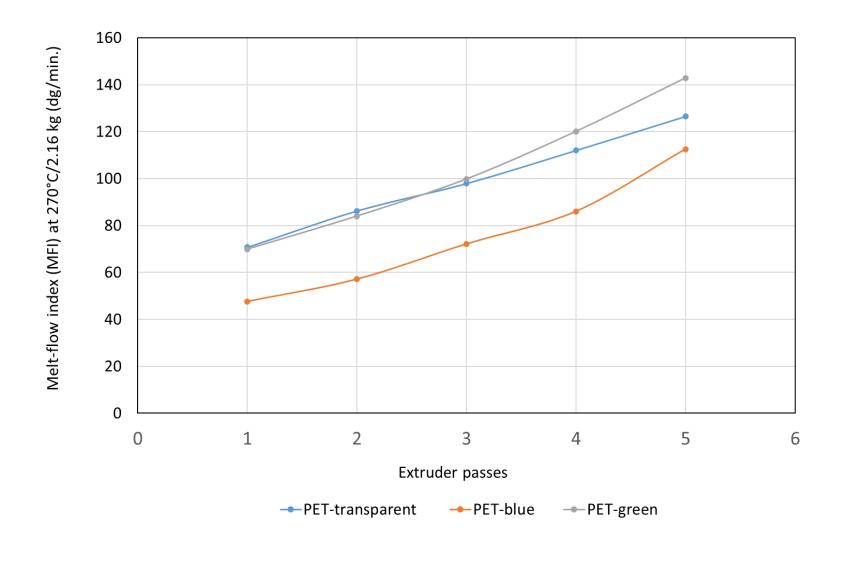
Before and between extrusions dried 3 hrs at 150 °C

Melt density 1.2 g/cm³ used for MFI calculation





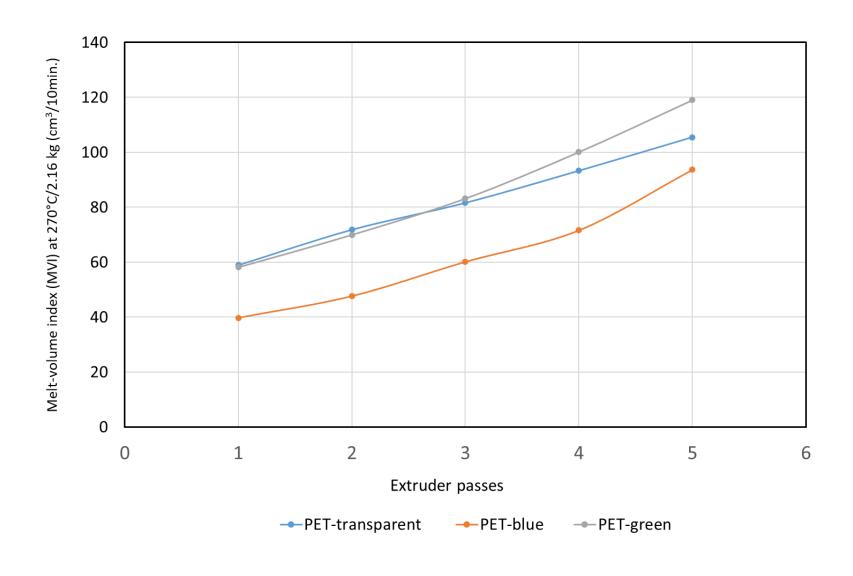




Comparison PET

Processing stability overall comparison (MFI - melt-flow index)





Comparison PET

Processing stability overall comparison (MVI - melt-volume index)

HDPE-caps

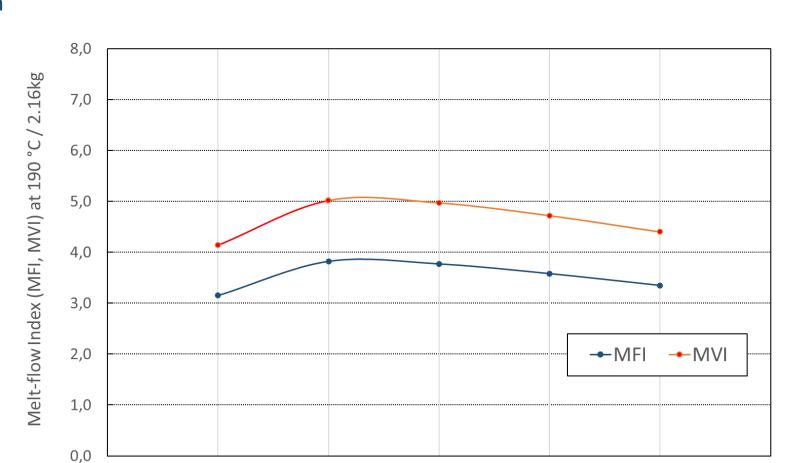
Processing stability determination

HAAKE single-screw extruder 19 mm, L/D=25

5-fold extrusion at 240°C/100 rpm

No drying before and between extrusions

Melt density 0.76 g/cm³ used for MFI calculation



3

Exruder passes

4

6

2

1

0

HDPE-mix

CEVOOH

Processing stability determination

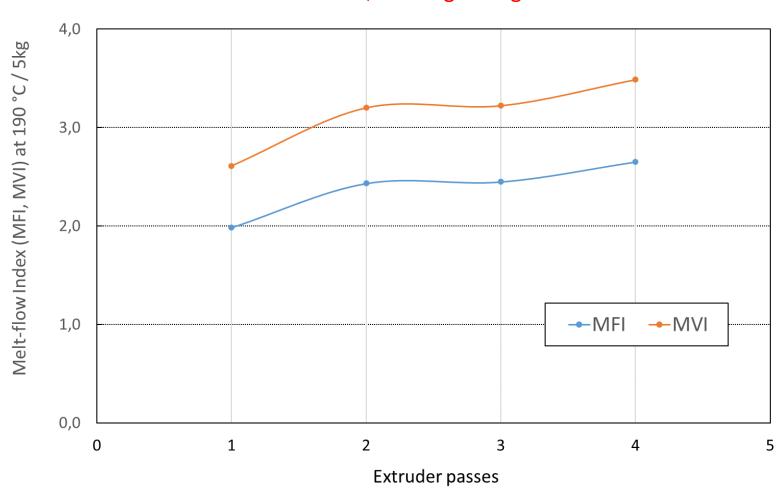
HAAKE single-screw extruder 19 mm, L/D=25

4-fold extrusion at 240°C/100 rpm

No drying before and between extrusions

Melt density 0.76 g/cm³ used for MFI calculation

Reference material, not originating from drink bottles



PP-labels



Processing stability test

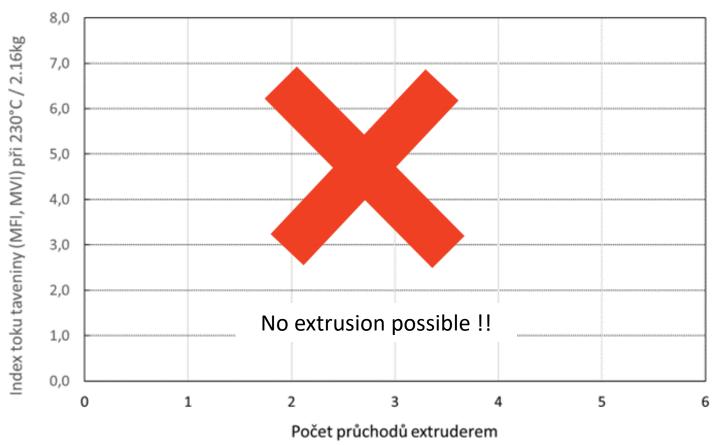
HAAKE single-screw extruder 19 mm, L/D=25

1-fold extrusion at 220°C/100 rpm

No drying before extrusion

Observations:

Extremely low viscosity Inhomogeneous melt High portion of volatiles Strong smell and smoke



High surface/mass ratio and an extremely high portion of print colours on the film surface make the polymer label material unprocessable, i.e. useless (!)

Conclusion (1) Role of laboratory testing



- Laboratory tests are the first step to the assessment of possible plastic waste re-processing.
- Laboratory tests have technical draw-backs that have to be taken into account when looking for the correlation with the plant processing line.
- Despite some specific features, **laboratory tests are irreplaceable both from the technical and economic reasons**. They will reveal if recycling the given material makes sense or not (!).

Conclusion (2) Recycling and plastics life-time period



- Direct recycling of any plastic always provides material of second-rate quality.
- Direct recycling does not solve the end of plastic life. It only postpones it (!).
- Presently, the real and most efficient elimination of plastic waste is reached only by Waste-to-Energy.
- Chemical recycling can be the potential future solution (!).

